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July 21, 2000

Commissioner of Patents  
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United States Patent Application  
METHOD FOR THE TREATMENT OF A SOIL  
CONTAINING SOILBORNE PATHOGENS  
George Lazarovits - Inventor

Sir:

In order to effect the filing of the above noted United States Patent Application, enclosed herewith are the following:

- x Disclosure comprising pages 1 - 20;  
Claims 1 to 19 comprising pages 21 - 23; and  
Abstract of the Disclosure comprising page 24
- x Drawings Figures 1 to 15 comprising 13 pages
- x Government fees: Filing \$345.00 U.S.

The particulars for filing the above noted application are as follows:

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2.

Docket No. 2527-1A  
George Lazarovits


Title: METHOD FOR THE TREATMENT OF A SOIL  
CONTAINING SOILBORNE PATHOGENS

Please note that the undersigned, Eric Fincham, Reg. 28,201, is the Agent of Record for the enclosed Patent Application. Correspondence may be addressed to:

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It is believed that the enclosed papers are all in order for filing and in due course, we look forward to receiving the Official Filing Receipt.

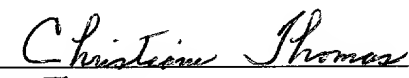
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Date *July 24, 2000*

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## METHOD FOR THE TREATMENT OF A SOIL

### CONTAINING SOILBORNE PATHOGENS

#### FIELD OF THE INVENTION

The present invention relates to the control of soilborne pathogens and to a method for determining the treatment of a soil containing soilborne pathogens.

#### BACKGROUND OF THE INVENTION

The treatment of soil in agricultural systems is widely practiced and essential since the growing of plant life remains a basic requirement for feeding the world population. Over the years, many different types of soil treatments have been used and experimented with depending on the desired result. Thus, it is common and widely known to use fertilizers to provide the nutrients for plant growth.

A common problem associated with most crops is the presence of soil pathogens and weeds within the soil. These soil pathogens substantially reduce the yields of any given crop and accordingly, many different soil treatments have been developed in order to rid the soil of such pathogens.

One of the more common and widely used methods of treating soils to eliminate pathogens therein is by the use of a methyl bromide. However, methyl bromide has been recognized as an ozone depleting chemical and as such, international agreement has stated that all production in developed countries must be phased out by the year 2005.

It has been estimated that the ban on methyl bromide will have a serious effect on crop damages and yields. In particular, crops such as potatoes, tomatoes, peppers,

strawberries, etc are particularly susceptible to soilborne pathogens.

While other products for treating soil have been proposed, they have not received any wide degree of acceptance. Other methods of controlling soil pathogens include crop rotation and field fallowing. However, these approaches lower the return on a given parcel of land.

#### SUMMARY OF THE INVENTION

It is therefore an object of the present invention to provide a method for controlling soil pathogens.

It is a further object of the present invention to provide a method to determine an effective treatment for soil pathogens.

According to one aspect of the present invention, there is provided a method of controlling soilborne pathogens in a soil which comprises the step of adding a nitrogen containing material and a pH reducing agent to the soil, the pH reducing agent being present in an amount sufficient to reduce soil pH below 5.5.

In a further aspect of the present invention, there is provided a method of determining an effective method for control of soil pathogens in a soil comprising the steps of measuring the pH of the soil, measuring the organic carbon content of the soil, measuring the buffering capacity of the soil, adding a nitrogen containing material and a pH reducing agent to reduce soil pH below 5.5 when the buffering capacity is below 2 uL H<sub>2</sub>SO<sub>4</sub> g/soil, and adding a nitrogen containing material at a pH raising agent to raise the pH above 8.5 when the organic carbon content is less than 1.7% by weight.

In a further aspect of the present invention, there is provided a method of controlling soilborne pathogens in a soil having an organic carbon content less than 1.7% by weight comprising the step of adding a nitrogen containing material and a pH raising agent to raise soil pH above 8.5.

It is being found that the most effective method of controlling soil pathogens will depend upon key properties of the soil. In particular, it is being found that soilborne pathogens can be controlled by exposing the pathogens to either ammonia or nitrous acid in a sufficient concentration, one method being selected over the other depending upon soil properties.

In particular, it is known that ammonia is effective for the control of soilborne pathogens. However, it has been found that the addition of even large amounts of a nitrogen containing compound sufficient to generate ammonia will not function when the pH of the soil is at below critical level and when the organic carbon content of the soil is greater than 1.7% on a weight basis.

Similarly, it has been found that nitrous acid ( $\text{HN0}_2$ ) and/or nitrite ( $\text{N0}_2$ ) is effective when the pH of the soil is reduced to a level below 5.5 and preferably below 5.

The fungus *Verticillium dahliae* Kleb., a wilt pathogen of many crops, is used as a model pathogen. In potato (*Solanum tuberosum*), *Verticillium* wilt causes premature senescence and the disease is aptly referred to as "early dying syndrome". Infection of potato plants occurs when roots contact microsclerotia (MS) of *V. dahliae*. Microsclerotia over winter in soil and consist of clustered, melanized,

thick-walled and hyaline, thin-walled hyphal cells. *Verticillium* wilt is difficult to manage because of the limited success of crop rotations, the slow development of resistant cultivars and the absence of chemical control options. Our studies and those of others have shown that when one controls this organism in the soil, then many other pathogenic agents, nematodes, weeds and pests are similarly controlled.

The following examples and figures exemplify different aspects of the invention, wherein;

Figures 1a to 1k are graphs indicating soil pH and the number of microsclerotia germinated as well as  $\text{NH}_3$  concentration in soil,  $\text{NO}_2^-$  and  $\text{NO}_3^-$  content in two different soils;

Figures 2a through 2e are graphs showing microsclerotia germinated in a single soil;

Figures 3a through 3h are graphs showing the effect with and without the nitrification inhibitor DCD;

Figures 4a through 4d are graphs showing the number of microsclerotia germinated and  $\text{HNO}_2$  concentration;

Figures 5a through 5m are graphs similar to those in Figure 3 in a different type of soil;

Figure 6 is a graph illustrating the number of microsclerotia germinated after being exposed for two weeks to various concentrations of  $\text{NH}_3$ ;

Figure 7 is a graph illustrating the number of microsclerotia germinated for various time counts after exposure to various concentrations of  $\text{NH}_3$ ;

Figure 8 is a graph illustrating the number of microsclerotia germinated at various times after being exposed to various concentrations of  $\text{HNO}_2$  and a citric acid buffer;

Figure 9 is a graph illustrating the number of microsclerotia germinated after exposure to various concentrations of 30 mL  $\text{HNO}_2$  and a citric acid buffer;

Figure 10 is a graph illustrating the peak concentration of  $\text{NH}_3$  for a soil amended with 2% MBM;

Figure 11 is a graph illustrating soil pH in response to  $\text{H}_2\text{SO}_4$ ;

Figures 12a through 12k are graphs illustrating the number of microsclerotia germinated, soil pH,  $\text{NO}_2^-$  and  $\text{NO}_3^-$  content and  $\text{HNO}_2$  concentration in a soil solution and two different soils;

Figures 14a, 14b, and 14c are graphs illustrating the germination of microsclerotia and percent colony forming units of different organisms.

The viability of *V. dahliae* microsclerotia retrieved from soil was determined using the following experimental model system. Desired levels of nitrogenous product comprising meat and bone meal (MBM) were mixed with soil and 20 g of the mixture added to 50 mL test tubes. Microsclerotia suspended in crushed silica sand was added to a nylon pouch which was then buried in the soil or suspended in the head space of each tube. Water was added to each tube to bring soil moisture to 0.33 bar tension. The tubes were then loosely capped, and placed in the dark at 24°C. On each date of analysis, the nylon pouches were retrieved and an Anderson air sampler used to impact the pouch contents onto a medium (soil-pectate-tergitol)

selective for growth of *V. dahliae*. Viability of microsclerotia were determined two weeks following plating as the number of microsclerotia germinated from a total of 50 examined. Resulting death of microsclerotia using this bioassay has been shown to correlate to the reduction of wilt disease incidence in greenhouse and field planted potatoes.

The levels of  $\text{NH}_3$ ,  $\text{NO}_2^-$  and  $\text{NO}_3^-$  in soil were determined at each sample date. Soil (8g) was mixed with cold distilled water (40mL) in sealed plastic bags, the slurry mechanically disrupted with a Stomacher laboratory blender, and shaken at 5°C for one hour. The slurry was once again mechanically disrupted and its pH determined. The slurry was centrifuged and supernatant analyzed for total ( $\text{NH}_3 + \text{NH}_4^+$ ), ( $\text{HNO}_2 + \text{NO}_2^-$ ) and  $\text{NO}_3^-$  using an ion chromatograph. Ammonia and  $\text{HNO}_2$  were calculated as the fraction of total ( $\text{NH}_3 + \text{NH}_4^+$ ) or ( $\text{NO}_2^- + \text{HNO}_2$ ) respectively in solution using the Henderson-Hasselbalch equation knowing soil pH and incubation temperature.

Ammonia in excess of 65 mg N kg<sup>-1</sup> soil (20 mM  $\text{NH}_3$ ) coincided with a rapid loss in the viability of microsclerotia (**Fig. 1**). In two experiments MBM or soya meal (SM) were added to various concentrations (0, 0.25, 0.5, 1, and 2% weight/weight) to soils from two locations namely, Beauseart and Thorndale. Quite high levels of ammonia accumulated in the Beauseart soil amended to 2%, but none was detected in the Thorndale soil. The viability of microsclerotia remained above 60% in Thorndale soil compared to less than 10% in Beauseart soil amended to 2% (weight/weight). When 1% MBM or SM was added to Beauseart soil a gradual decline in



microsclerotia viability to 0% was seen four weeks after amendment. The decline in microsclerotia viability coincided with decreasing soil pH from 8 to 6 and  $\text{NO}_3^-$  accumulation in soil. Ammonia accumulation was negligible in Beauseart soil amended to 1% MBM (weight/weight), suggesting it was not involved in the death of microsclerotia at this rate. The results suggest the microsclerotia were killed by acute  $\text{NH}_3$  toxicity in Beauseart soil amended to 2% (weight/weight) with MBM or SM and by quite a different mechanism when amended to 1%.

The Thorndale soil amended to 2% MBM or SM failed to accumulate sufficient  $\text{NH}_3$  to kill microsclerotia. This provided the opportunity to confirm  $\text{NH}_3$  as responsible for killing of microsclerotia by inducing high levels of  $\text{NH}_3$  in the Thorndale soil by determining the survival of microsclerotia. This approach consisted of adding high rates of MBM to the Thorndale soil. Thus MBM was applied at the rates of 0,2 and 4% (weight/weight). The 2% amendment resulted in negligible  $\text{NH}_3$  accumulation and survival of microsclerotia greater than 50% by the end of the study (Fig. 2). In contrast at 4% MBM,  $\text{NH}_3$  accumulated to above 150 mM one week following amendment and continued to the end of the study. This corresponded to complete death of microsclerotia.

Accumulation of greater than 10 mM  $\text{NH}_3$  was consistently accompanied with a rapid decline in the viability of microsclerotia in soil. Based on this observation the toxicity of  $\text{NH}_3$  to microsclerotia in solution and atmosphere was tested and compared to levels required in soil to kill microsclerotia in soil.

Germination of microsclerotia was prevented by  $\text{NH}_3$  concentrations larger

than 3 mM in agar medium (**Fig. 7**).  $\text{NH}_3$  was generated in SPT medium by addition of various concentrations of  $\text{NH}_4\text{Cl}$  to the medium and varying the pH of the medium (7, 7.6, 8 and 8.5). Immediately following cooling and hardening of the medium, 25 microsclerotia per Petri dish were transferred individually with a needle and germination recorded two weeks later. Microsclerotia that failed to germinate were transferred to regular SPT (containing no  $\text{NH}_3$ ) and still did not germinate.

Various concentrations of  $\text{NH}_3$  were generated in glycine or tricine buffer solution at pH 8.6 with  $\text{NH}_4\text{Cl}$  or  $(\text{NH}_4)_2\text{SO}_4$  added. A 15mL test tube was filled with appropriate buffer and  $\text{NH}_4^+$  solution, microsclerotia added, the tube capped, then placed in the dark at  $24^\circ\text{C}$  and the tubes inverted daily to suspend the microsclerotia in solution. Microsclerotia survival was determined by emptying contents of the tube into a Buchner funnel, the microsclerotia being retained on Whatman #42 filter paper, rinsed with distilled water, transferred to SPT medium by placing the filter paper in contact with the agar medium and removed leaving the microsclerotia adhering to the medium.

Microsclerotia germination decreased with concentration and duration of exposure to  $\text{NH}_3$  in glycine buffer (**Fig. 8**). An exposure of four days to greater than 5 mM  $\text{NH}_3$  prevented germination of microsclerotia. Germination of microsclerotia to various concentrations of  $\text{NH}_3$  was not affected by buffer (glycine or tricine),  $\text{NH}_4^+$  source ( $\text{NH}_4\text{Cl}$  or  $(\text{NH}_4)_2\text{SO}_4$ ) or NaCl concentrations equivalent to the N sources added.

In previous experiments,  $\text{NH}_3$  was found to accumulate in the Thorndale but

not Bearseart soil amended to 2% MBM (weight/weight). Therefore a series of experiments were conducted to determine the soil properties preventing  $\text{NH}_3$  accumulation and microsclerotia death in soil.

Several studies have reported that soil organic matter or clay can absorb  $\text{NH}_3$  to exchangeable negatively charged sites. Thus various amounts of  $\text{NH}_4\text{OH}$  were added to either soil (varying in levels of  $\text{NH}_3$  eq. to 0 to 4% MBM weight/weight), soil brought to 0.333 bar, incubated for four days with subsequent extraction for estimation of  $\text{NH}_3$ . Addition of about 1800 mg  $\text{NH}_4\text{OH-N kg}^{-1}$  (eq. to 2 MBM weight/weight) to the Thorndale or Beauseart soil was sufficient to induce levels of  $\text{NH}_3$  sufficient to kill microsclerotia (data not shown). Therefore, it seems retention of  $\text{NH}_3$  in amended Thorndale soil cannot explain the occurrence of insufficient  $\text{NH}_3$  levels to kill microsclerotia.

Rapid nitrification was observed in the Thorndale soil amended to 2% MBM. To test if this rapid nitrification converted  $\text{NH}_3$  to  $\text{NO}_2^-$  and  $\text{NO}_3^-$ , thus preventing  $\text{NH}_3$  accumulation, the nitrification inhibitor dicyandiamide (DCD) was added with MBM to the Thorndale soil. Addition of inhibitor prevented the accumulation of  $\text{NO}_2^-$  or  $\text{NO}_3^-$  and the reduction in soil pH following amendment to 2% MBM. However,  $\text{NH}_3$  levels in soil were not sufficient to kill microsclerotia, thus nitrification alone cannot be attributed to prevention of  $\text{NH}_3$  toxicity in the Thorndale soil (data not shown).

To determine the factor(s) that control  $\text{NH}_3$  accumulation in soil, 2% MBM was added to twelve soils with a range of soil properties including texture, organic

carbon, cation exchange capacity,  $\text{NH}_3$  and acid buffering capacity. The Beauseart and Thorndale soils studied previously were included in the twelve soils. Selected soil properties including  $\text{NH}_3$ ,  $\text{NH}_4^+$ ,  $\text{NO}_2^-$ ,  $\text{NO}_3^-$ , pH, soil C:N ratio, electrical conductivity, total bacteria, total fungi, ammonifying bacteria, ammonifying fungi, proteolytic bacteria and soil respiration which may influence the accumulation of  $\text{NH}_3$  in soil were measured over time. Microsclerotia were killed within one week of addition in four of the twelve soils amended. An accumulation of  $\text{NH}_3$  (greater than  $65 \text{ mg N kg}^{-1}$  or  $20 \text{ mM}$ ) was found in each of these soils one week following amendment. Organic carbon content of soil was the only soil property highly related to  $\text{NH}_3$  accumulation ( $r=0,92$ ) with each of the four soils containing less than 1.7% organic carbon (weight/weight) (**Fig. 11**).

Ammonia failed to accumulate to toxic levels in soils with an organic carbon content larger than 1.7% amended to 2% MBM. To confirm the role of organic carbon in soils controlling the level of  $\text{NH}_3$  accumulation following amendment, a recalcitrant source of carbon (Holland Marsh Muck soil) was added to Beauseart and Habsor (sand) soils and amended with MBM. Amendment of the Beauseart soil to 2% MBM resulted in death of microsclerotia by day 9 with associated high levels of  $\text{NH}_3$  in soil. In contrast addition of Holland Marsh soil to 2 and 4% (weight/weight) together with MBM to 2% in Beausaert soil resulted in survival of microsclerotia with negligible levels of  $\text{NH}_3$  present in soil (data not shown). Addition of Holland Marsh soil to 5% (weight/weight) of the Habsor sand amended to 2% MBM (weight/weight) resulted in greater than 80% survival of microsclerotia with an insufficient amount of

NH<sub>3</sub> in soil (less than 15 mg NH<sub>3</sub>-N kg<sup>-1</sup> or 7 mM NH<sub>3</sub>) required to kill microsclerotia (data not shown).

The lack of NH<sub>3</sub> accumulation in the Thorndale soil amended with MBM or SM was attributed to soil pH remaining below 8.5, being insufficient to convert NH<sub>4</sub><sup>+</sup> to NH<sub>3</sub>. Calcium oxide (CaO) stabilized sewage sludge (pH 13) was added to soil to raise soil pH and induce NH<sub>3</sub> toxicity. The sludge raised soil pH above 8.5, during the first four days following its addition. Only when the sludge was added five days following MBM amendment were microsclerotia killed (**Table 1**). The NH<sub>4</sub><sup>+</sup> released during decomposition and mineralization of MBM at day 5 was converted to NH<sub>3</sub> by the liming effect of the CaO stabilized sludge, thus NH<sub>3</sub> toxicity induced.

**Table 1. Number of microsclerotia germinated out of 50 counted with 0 or 2% added and 0 or 4% CaO stabilized municipal sewage sludge added on day 0 or day 5 after M.M. addition.**

% MBM (weight/weight)	% Sludge (weight/weight)	Day of Sludge Addition	MS germination (N=3 standard error of the mean)
0	0	-	50 (0.33)
0	4	0	48 (1.15)
2	0	-	45 (0)
2	4	0	44 (1.53)
0	0	-	46 (0.33)
0	4	5	42 (2.19)
2	0	-	47 (1.20)
2	4	5	0 (0)

Base generating agents such as calcium hydroxide, calcium oxide, sodium hydroxide, ammonium hydroxide, and potassium hydroxide can be added to various soils to increase soil pH and encourage the generation of  $\text{NH}_3$  from N amendments. By doing so the required rate of N amendment to disinfect soil of soilborne pathogens will be reduced to economical and environmentally suitable levels. Further, the amount of the base agents required to bring soil pH to desired levels to induce generation of  $\text{NH}_3$  can be determined. Soil properties such as organic matter content and initial soil pH are being used to predict the amount of base agent and N amendment required to disinfect soil of plant pathogens and pests.

The germination of *V. dahliae* MS having been exposed to varying concentrations of  $\text{NH}_3$  was determined. *V. dahliae* MS were exposed to  $\text{NH}_3$  in solid agar medium, in buffered solutions, and in the atmosphere above buffered solutions containing  $\text{NH}_3$ . All toxicity studies were done in triplicate and repeated once.

**Agar medium:** Various amountes of  $\text{NH}_4\text{Cl}$  salt (to 0, 25, 50, 100 and 200 mM) was added to soil peclate medium (SPT) and the pH of the medium adjusted to 7.0, 7.6, 8.0 or 8.5 by addition of 5 M NaOH. For each level of  $\text{NH}_4\text{Cl}$  and pH, three replicate dishes were poured. The agar was allowed to solidify for 2 hours and immediately thereafter 25 *V. dahliae* MS were individually transferred to each dish using a sterile hypodermic needle. The dishes were wrapped with Parafilm (American National Can, Neenah WI) to limit loss of  $\text{NH}_3$  by volatilization, and incubated for two weeks at 24°C in the dark. The viability of *V. dahliae* MS was determined as percent of total MS forming colonies. The concentration of  $\text{NH}_3$  in medium was

estimated as described below and ranged from 0 to 31 mM. The pH of medium after hardening and two weeks after incubation was tested using pH test strips (ColorpHast pH 5-10; EM Science, Gibbstown NJ) and was within 0.5 units (limit of resolution of the test strips) of that set at the start. The results are shown in **Fig. 3(a)**.

**Buffered solutions:** Varying amounts of a 2.70 M  $\text{NH}_4\text{Cl}$  stock solution was added (from 0 to 1.0 mL) to 40 mL of 0.05 M glycine solution adjusted to pH 8.6 with NaOH (Perrin & Dempsey 1974). The final volume was brought to 50mL with the same glycine solution. Solutions were then sterilized by filtration through a 0.22  $\mu\text{m}$  pore size filter and 15 mL placed into three replicate sterile screw cap tubes (total capacity 15.5 mL) About 200 *V. dahliae* MS were immediately added to each tube and the tube closed and placed at 24°C and in the dark. Tubes were inverted every 12 hours to mix and suspend *V. dahliae* MS in the solution. A maximum duration of exposure of four days was chosen because *V. dahliae* MS germinated after five days in solutions of 0 to 0.65 mM  $\text{NH}_3$ . At 8 hours, 1 and 4 days, the contents of each tube was passed onto a Buchner funnel containing sterile filter paper (Whatman #42). The filter paper retained the *V. dahliae* MS and was rinsed with sterile water and placed onto the surface of SPT medium such that *V. dahliae* MS contacted the agar surface. The paper was then removed, leaving *V. dahliae* MS adhering to the surface of the medium. The viability of *V. dahliae* MS was determined as the percentage of MS forming colonies out of 50 counted per replicate dish. The concentration of  $\text{NH}_3$  in solutions was estimated as described below. The effect of type of buffer solution or  $\text{NH}_4^+$  salt on survival of *V. dahliae* MS was tested by adding

(NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> to glycine buffer solutions instead of NH<sub>4</sub>Cl, NaCl instead of NH<sub>4</sub>Cl to glycine buffered solutions, and NH<sub>4</sub>Cl added to tricine buffered solutions. The solution concentration of NH<sub>4</sub><sup>+</sup>+NH<sub>3</sub> in tubes after 4 days was within 5% of that measured at the start of the assay. Solution pH after 4 days did not vary from the set pH at the start of the assay by more than 0.1 units. The results are shown in **Fig. 3(b)**.

**Ammonia gas:** Fifty mL of a prepared solutions of NH<sub>4</sub>Cl in glycine solution described previously were added to sealer jars (each 250 mL). A mesh bag containing *V. dahliae* MS was then suspended in the atmosphere of the jar using a paper clip attached to a septum fitted in the lid of the jar. The jar was sealed, and placed at 24°C in the dark for 4 days. The mesh bags were retrieved and the viability of *V. dahliae* MS determined.

The survival of the other test organisms exposed to NH<sub>3</sub> was determined in glycine buffered solutions prepared as described previously. Sclerotia of *S. sclerotiorum* (15 per treatment), seeds of *A. retroflexus*, *L. sativa* and *R. sativus* (50 to 100 per treatment) were added to each tube containing a test solution. For *S. scabies* and FOL a one mL suspension of spores and chlamydospores respectively were added to each tube containing 14 mL of concentrated NH<sub>4</sub>Cl/glycine solution. The propagule density in the added suspension was prepared in distilled water and adjusted to give about 50 colony forming units (cfu) per 0.1 mL of test solution at the start of the experiment. Test solutions containing *S. scabies* and FOL were placed on YME and PDA medium respectively. Three replicate platings were made for each test solution of *S. scabies* and FOL and the cfu count were averaged. The viability of



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sclerotia of *S. sclerotiorum* was determined as colony formation (non-carpogenic germination) on PDA medium and that of seeds on water agar (WA) medium (1.5% agar) in Petri dishes. Sclerotia and seeds were separated from test solutions in a similar manner as that for *V. dahliae* using a Buchner funnel and sterile filter paper. Sclerotia were cut in two, and a total of 5 placed (cut surface down) onto a dish containing medium. Seeds were transferred to WA medium directly from the filter paper as done for *V. dahliae*. Dishes were immediately wrapped using stretchable sealer tape. All dishes containing propagules of the test organisms were placed at 24°C in the dark. Seed survival was based on the development of a 2 mm radicle. Seeds of *R. sativus* that failed to germinate after exposure to NH<sub>3</sub>, were checked for viability by staining with tetrazolium salt (2:3:5-triphenyl-tetrazolium chloride; BDH, Poole UK) according the procedure outlined by Moore (1973). The viability of all organisms tested was expressed as a percentage of that determined for the 0 mM test solution at the start of the experiment. For each test organism, this experiment was done in triplicate and repeated once. The results are shown in Fig. 3(c).

The germination of *V. dahliae* MS was determined using a setup similar to that described above for exposure of *V. dahliae* MS to NH<sub>3</sub> in buffered solution. Exceptions being various amounts of 0.270 M NaNO<sub>2</sub> stock solution were added (from 0 to 2.0 mL) to 40 mL of 0.02 M citric acid solution (set to pH 5.0 with NaOH (Perrin and Dempsey 1974)) and brought to 50 mL volume with the same citric acid solution. All experiments were done in triplicate and repeated. Solutions were filter sterilized and 15 mL placed into sterile screw cap tubes (total capacity 15.5 mL).

Immediately, *V. dahliae* MS were added to each tube, the tube closed, placed at 24°C and in the dark and inverted twice daily. At 8 hours, 1 and 4 days, *V. dahliae* MS viability was determined. The concentration of HNO<sub>2</sub> in solutions was estimated as described below. The effect of NO<sub>2</sub><sup>-</sup> on survival of *V. dahliae* MS was tested by adding various amounts of NaNO<sub>2</sub> (0 to 53 mM NO<sub>2</sub><sup>-</sup>) to citric acid buffer and pH set to 4.0, 5.0 and 6.0. The effect of type of NO<sub>2</sub><sup>-</sup> salt on survival of *V. dahliae* MS was tested by adding KNO<sub>2</sub> or NaCl instead of NaNO<sub>2</sub> to citric acid buffer (set to pH 5.0). The concentration of NO<sub>2</sub><sup>-</sup> + HNO<sub>2</sub> and pH of solutions after 4 days was within 5% and 0.1 units of the solution at day 0, respectively. The results are shown in **Fig. 14(a)**.

Exposure of *V. dahliae* MS to HNO<sub>2</sub> gas was done using a similar procedure described above. To sealer jars (250 mL), 50 mL of a prepared solution of NaNO<sub>2</sub> in citric acid solution was added. A *V. dahliae* MS bag was suspended in the atmosphere of the jar. The jar was sealed, and placed at 24°C in the dark for 4 days. Thereafter the bags were retrieved and the viability of *V. dahliae* MS determined. This experiment was repeated and done in triplicate. The results are shown in **Fig. 14(b)**.

Determination of the survival of the other test organisms exposed to HNO<sub>2</sub> was done using citric acid buffered solutions as described above and detailed by Tenuta and Lazarovits (2000a). Sclerotia of *S. sclerotiorum*, chlamydospores of FOL, spores of *S. scabies*, seeds of *A. retroflexus*, *L. sativa* and *R. sativus* were added to each tube containing 15 mL of HNO<sub>2</sub>, in citric acid solution at pH 5.0. Tubes containing seeds

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were opened for 15 s on day 2, as a precaution to prevent generation of anaerobic conditions from respiration of seeds. The solution assay was done twice and in triplicate for each organism tested.

Seeds of *R. sativus* that failed to germinate after exposure to  $\text{HNO}_2$ , were checked for their viability by staining with tetrazolium salt (2:3:5-triphenyl-tetrazolium chloride; BDH, Poole UK) according the procedure outlined by Moore (1973). The germination or cfu of all organisms tested was expressed as a percentage of the germination or cfu of a control solution (0 mM  $\text{HNO}_2$ ) at the start of the experiment. The results are shown in **Fig. 14(c)**.

Various concentrations of  $\text{HNO}_2$  were generated in citric acid buffer solution at pH 4.0, 5.0 or 6.0 with  $\text{NaNO}_2$  added. A 15mL test tube was filled with appropriate buffer and  $\text{NaNO}_2$  solution, microsclerotia added, the tube capped, placed in the dark at 24°C and tubes inverted daily to suspend microsclerotia in solution. Microsclerotia survival was determined as described previously for  $\text{NH}_3$  in glycine buffer. Microsclerotia survival decreased with increasing concentration of  $\text{NaNO}_2$  and decreasing pH at a 24 hour exposure. A calculated concentration of above 0.10 mM  $\text{HNO}_2$  was required to kill all microsclerotia (data not shown). Further, the survival of microsclerotia was dependent upon the duration of exposure to  $\text{HNO}_2$ , about 0.025 mM  $\text{HNO}_2$  was sufficient to kill all microsclerotia at a four day exposure (**Fig. 9**). This is within the range of critical  $\text{HNO}_2$  concentration required in soil to kill microsclerotia.

In studies described here, microsclerotia died when suspended above soil

amended with MBM, SM, and various fertilizers. Therefore,  $\text{HNO}_2$  was suspected to also kill microsclerotia in atmosphere. Various amounts of  $\text{HNO}_2$  in atmosphere were subjected to microsclerotia by suspending microsclerotia in sealer jars containing 30 mL of citric acid  $\text{HNO}_2$  buffer solution and incubated in the dark at room temperature for four days. Microsclerotia died when suspended in atmosphere above the 0.10 mM  $\text{HNO}_2$  solution (**Fig. 10**).

This finding is important because it demonstrates that both  $\text{NH}_3$  and  $\text{HNO}_2$  can kill microsclerotia through exposure in soil solution or soil atmosphere. Since in soil microsclerotia may reside in both, solution and atmosphere, either compound has the potential to kill microsclerotia.

Observations from the experiments described here indicate three factors determine nitrous acid toxicity in soil. They being: a) amendment rate, b) rapid nitrification and c) poor soil acid buffering capacity. Nitrification determines nitrous acid toxicity because the intermediate  $\text{NO}_2^-$  is produced under conditions of rapid nitrification and the oxidation  $\text{NH}_4^+$  to  $\text{NO}_2^-$  generates protons which acidifies soil. The ability of a soil to buffer against the acidity generated during nitrification determines the relative amounts of  $\text{HNO}_2$  and  $\text{NO}_2^-$  according to soil pH. A acid buffering assay was developed in which various amounts of  $\text{H}_2\text{SO}_4$  were added to soil, incubated for two hours, distilled water added and the slurry shaken for one hour, the slurry then being allowed to settle for one hour with subsequent pH determination. Generally, soils group into two categories as may be seen in **Fig. 12**. Group 1 soils have the ability to accumulate  $\text{HNO}_2$  and thus toxicity to microsclerotia. Soils in this

group require less than 2 uL H<sub>2</sub>SO<sub>4</sub> g/soil to lower soil pH to 5. Those in Group 2 are soils in which greater than 6uL H<sub>2</sub>SO<sub>4</sub> g/soil weight is required to lower soil pH to 5. HNO<sub>2</sub> acid accumulation and toxicity to microsclerotia has not been demonstrated in soils belonging to this group. Soils in this group contain CaCO<sub>3</sub> this being the source of their buffering ability.

An example of the importance of nitrification rate in producing HNO<sub>2</sub> is evident in a study in which 400 or 800 mg N kg<sup>-1</sup> as (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> was added to Beauseart and Mackenzie soils. The Beauseart soil was air-dried and stored for 1.5 years prior to initiation of the experiment. The Mackenzie soil was recently collected and stored at 4°C and at field moisture content. Recently collected Beauseart soil was shown previously to generate HNO<sub>2</sub> in response to (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> addition (**Fig. 5**). However, the air-dried Beauseart soil failed to accumulate HNO<sub>2</sub> and kill microsclerotia (**Fig. 13**). In comparison, the Mackenzie soil has rapid nitrification, associated reduction in soil pH, accumulation of HNO<sub>2</sub>, and death of microsclerotia. The population of autotrophic nitrifying bacteria at the start of the experiment was higher in the Mackenzie soil ( $1.1 \times 10^5$  g<sup>-1</sup> soil) compared to the Beauseart soil ( $5.8 \times 10^3$  g<sup>-1</sup> soil) likely accounting for differences in nitrification rate between soils.

Many acid generating agents such as FeSO<sub>4</sub>, AlSO<sub>4</sub>, S<sup>0</sup>, SO<sub>2</sub>, H<sub>2</sub>SO<sub>4</sub>, ascorbic, sorbic, citric and acetic acids can be added to various soils to lower soil pH and encourage the generation of HNO<sub>2</sub> from N amendments. By doing so the required rate of N amendment to disinfest soil of soilborne pathogens is reduced to economical and environmentally suitable levels. The amount of the acid agents required to bring

soil pH to desired levels to induce generation of  $\text{HNO}_2$  can be determined. Soil properties such as  $\text{CaCO}_3$ , sand content and initial soil pH are used to predict the amount of acid agent and N amendment required to disinfest soil of plant pathogens and pests.

I CLAIM:

1. A method of controlling soilborne pathogens in a soil comprising the step of adding a nitrogen containing material and a pH reducing agent to said soil, said pH reducing agent being present in an amount sufficient to reduce soil pH below 5.5.
2. The method of claim 1 wherein said pH reducing agent is present in an amount sufficient to reduce soil pH below 5.0.
3. The method of claim 2 wherein said pH reducing agent is incorporated in said nitrogen containing material.
4. The method of claim 2 wherein said nitrogen containing material is selected from a group consisting of animal manures, sewage sludge, animal by-products, chitinous materials, oil- seed materials, urea,  $\text{NH}_4\text{x}$  and  $\text{xNO}_2$  compounds.
5. The method of claim 2 wherein said nitrogen containing material is added at a rate of between 200 kg N/ha and 1000 kg N/ha.
6. The method of claim 2 wherein said nitrogen containing material is an animal manure.
7. The method of claim 2 wherein said nitrogen containing material is a meat and bone meal material.
8. The method of claim 2 wherein said nitrogen containing material is soya meal.
9. The method of claim 5 wherein said nitrogen containing material is applied at a rate of between 400 kg N/ha and 800 kg N/ha.

10. The method of claim 7 wherein said nitrogen containing material is applied at a rate of between 600 kg N/ha and 1000 kg N/ha.

11. The method of claim 2 wherein said pH reducing agent is applied to said soil and said nitrogen containing material and subsequently added to said soil.

12. The method of claim 2 wherein said pH reducing agent is added to said soil subsequent to said nitrogen containing material.

13. The method of claim 2 wherein said pH reducing agent is selected from a group consisting of sulfuric acid, ascorbic acid, sorbic acid, citric acid, and  $\text{SO}_2$ .

14. The method of claim 2 wherein said pH reducing agent is an organic acid.

15. The method of claim 14 wherein said organic acid is citric acid.

16. A method of determining an effective method for control of soil pathogens in a soil comprising the steps of:

measuring the pH of a soil;

measuring the organic carbon content of said soil;

measuring the buffering capacity of said soil;

adding a nitrogen containing material and a pH reducing agent to reduce soil pH below 5.5 when said buffering capacity is below 2 uL  $\text{H}_2\text{SO}_4$  g/soil; and

adding a nitrogen containing material and a pH raising agent to raise said pH above 8.5 when said organic carbon content is less than 1.7% by weight.

17. A method of controlling soilborne pathogens in a soil having an organic carbon content less than 1.7% by weight, comprising the step of adding a nitrogen containing material and a pH raising agent to raise soil pH above 8.5.



18. The method of Claim 17 wherein said nitrogen containing material is selected from a group consisting of animal manures, sewage sludge, animal by-products, chitinous materials, oil- seed materials, urea,  $\text{NH}_4\text{x}$  and  $\text{xNO}_2$  compounds.
19. The method of Claim 17 wherein said nitrogen containing material is a meat and bone material.

## ABSTRACT OF THE DISCLOSURE

A method of determining an effective method for control of soil pathogens in a soil which includes the step of measuring the pH of the soil, measuring the organic carbon content of the soil, measuring the buffering capacity of the soil, adding a nitrogen containing material and a pH reducing agent to reduce the soil pH below 5.5 when the buffering capacity of the soil is below a predetermined amount, and adding a nitrogen containing material and a pH raising agent to raise the pH above 8.5 when the organic carbon content is less than 1.7% by weight.

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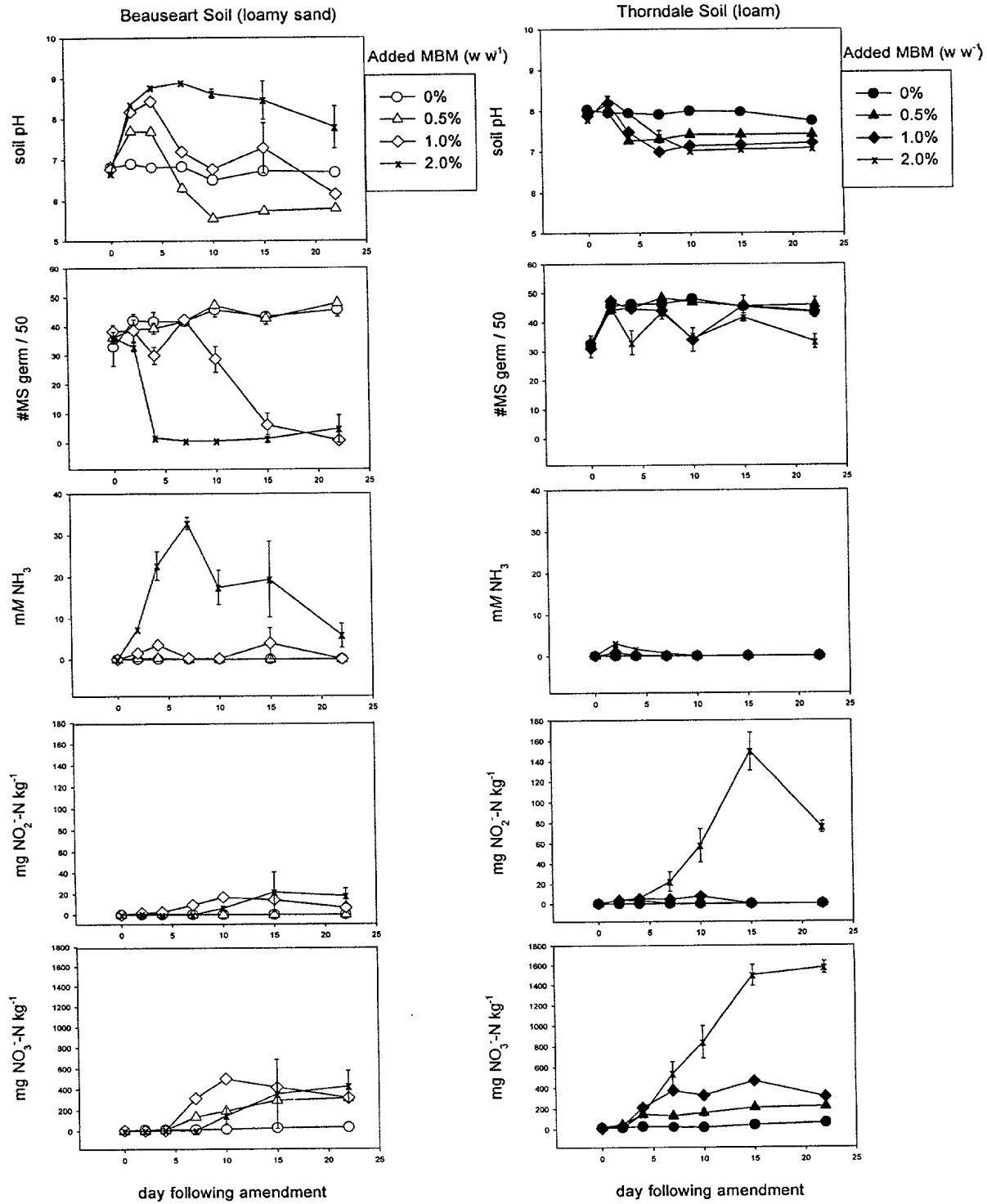


Fig. 1. Soil pH, # MS germinated out of 50 counted, NH<sub>3</sub> concentration in soil solution, NO<sub>2</sub><sup>-</sup> and NO<sub>3</sub><sup>-</sup> content in Beauseart and Thorndale soil amended with various rates of MBM (n=3; ±standard error).

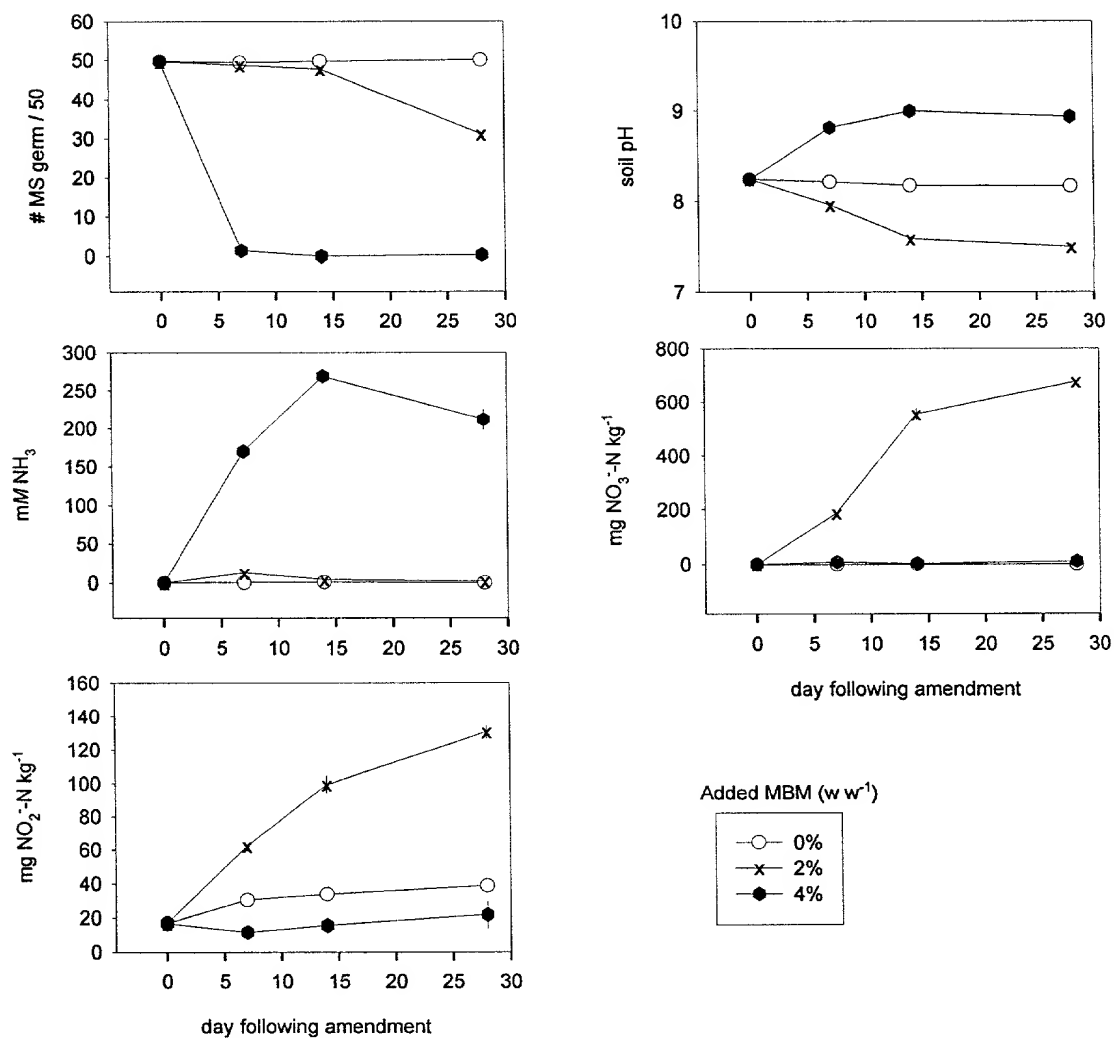


Fig. 2. Number MS germinated out of 50 counted, soil pH, NH<sub>3</sub> concentration in soil solution, NO<sub>3</sub><sup>-</sup> and NO<sub>2</sub><sup>-</sup> content of Thorndal soil amended with various rates of MBM (n=3; ±standard error).

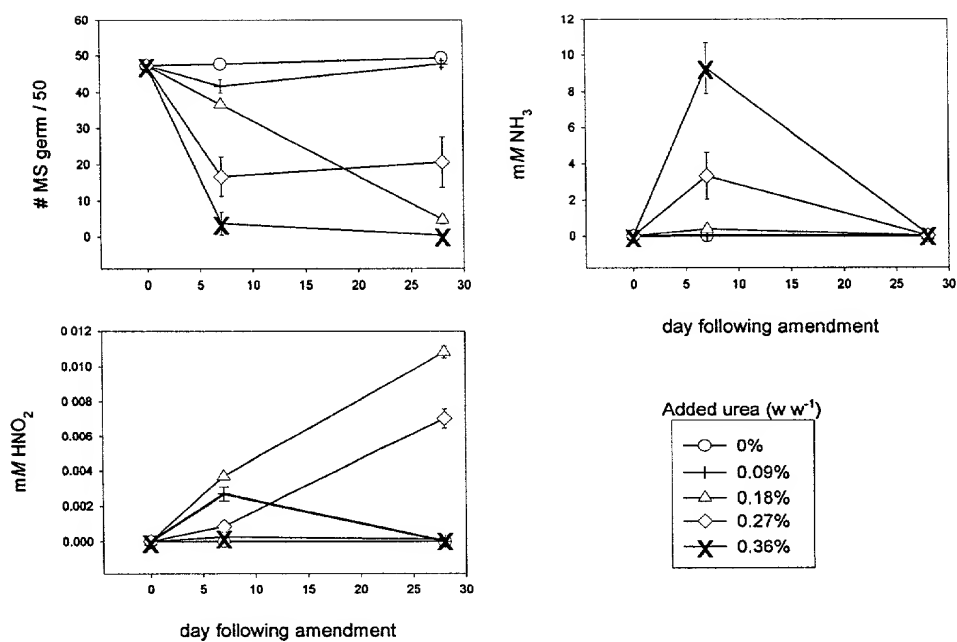


Fig. 3. Number MS germinated out of 50 counted, NH<sub>3</sub> and HNO<sub>2</sub> concentration in soil solution in Thorndale soil amended with various rates of urea (n=3;  $\pm$ standard error).

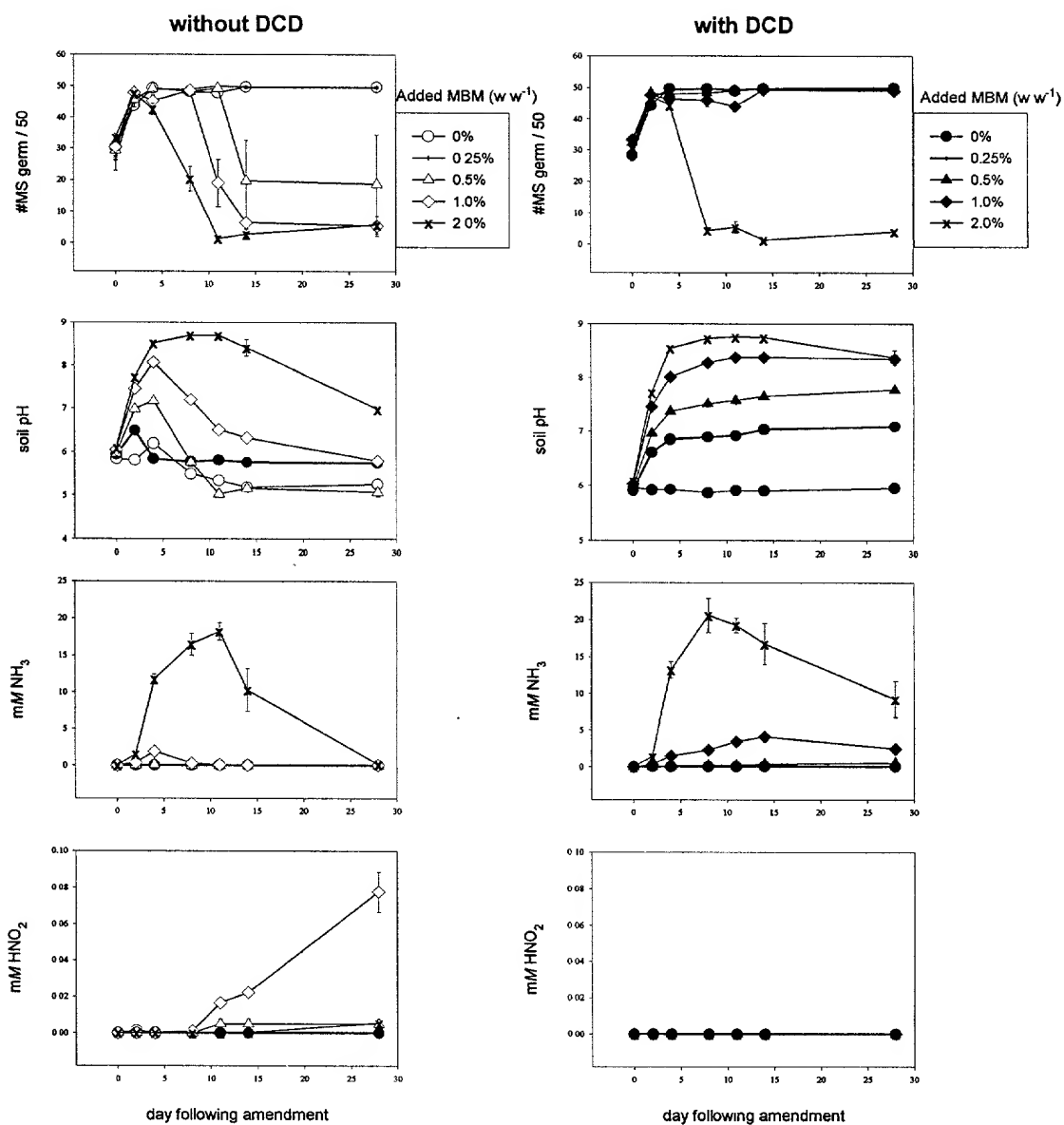


Fig. 4. Number MS germinated out of 50 counted, soil pH, NH<sub>3</sub> and HNO<sub>2</sub> concentration in soil solution in Beauseart soil (sandy loam) amended with various rates of MBM, with and without the nitrification inhibitor, dicyandiamide (DCD) added (n=3; ±standard error).

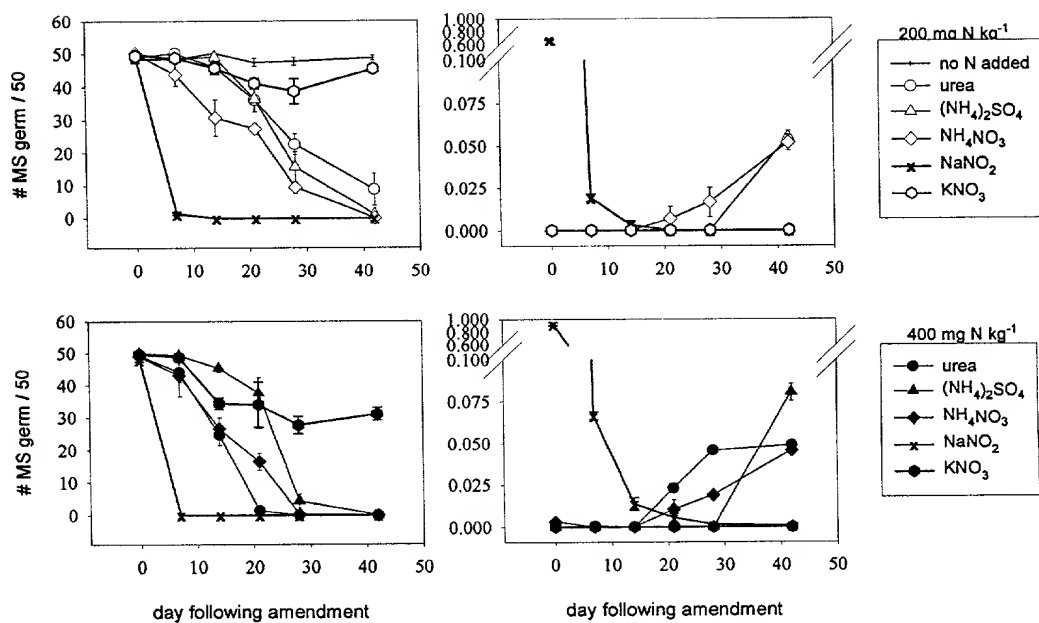


Fig. 5. Number of MS germinated out of 50 counted and HNO<sub>2</sub> concentration in soil solution of Beauseart soil amended with various fertilizer-N sources to 200 or 400 mg N kg<sup>-1</sup> (n=3; ±standard error).

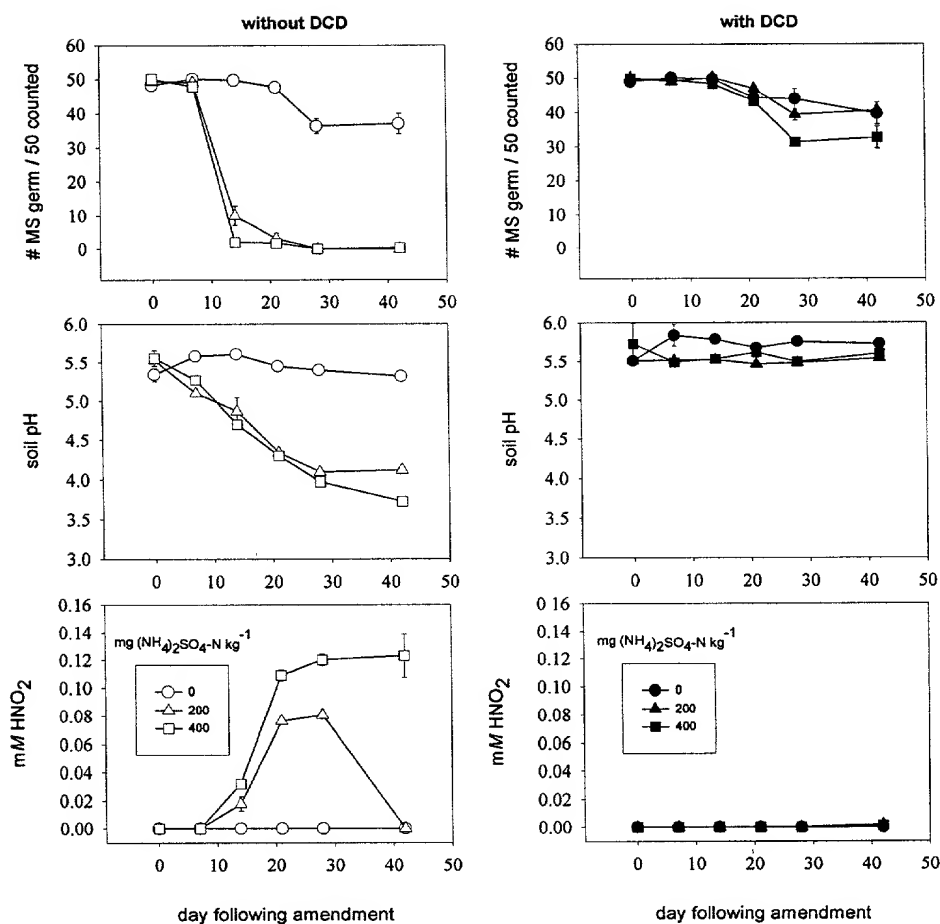


Fig. 6. Number of MS germinated out of 50 counted, soil pH and HNO<sub>2</sub> concentration in soil solution of Mackenzie soil amended with various amounts of (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>, with and without the nitrification inhibitor, dicyandiamide (DCD) added (n=3;  $\pm$  standard error).



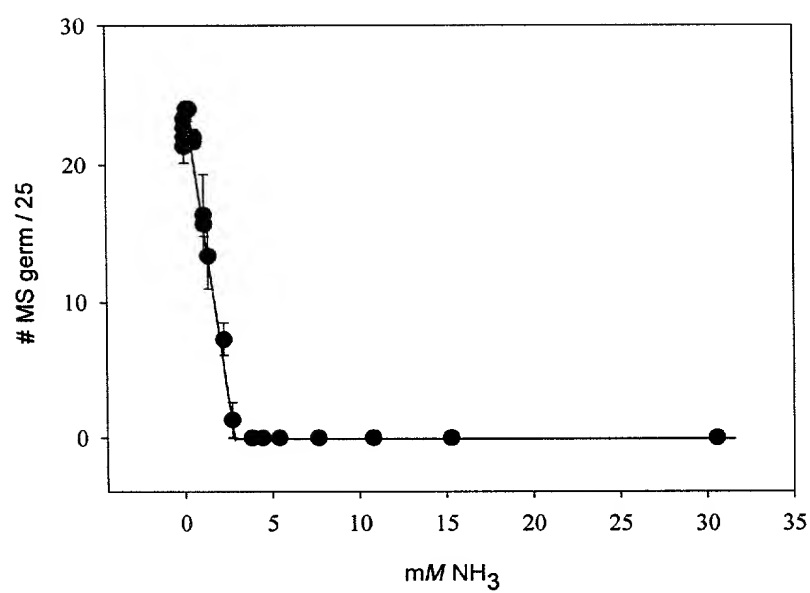


Fig. 7. Number of MS germinated out of 25 counted exposed for 2 weeks to various concentrations of NH<sub>3</sub> in solid medium (n=3;  $\pm$ standard error).

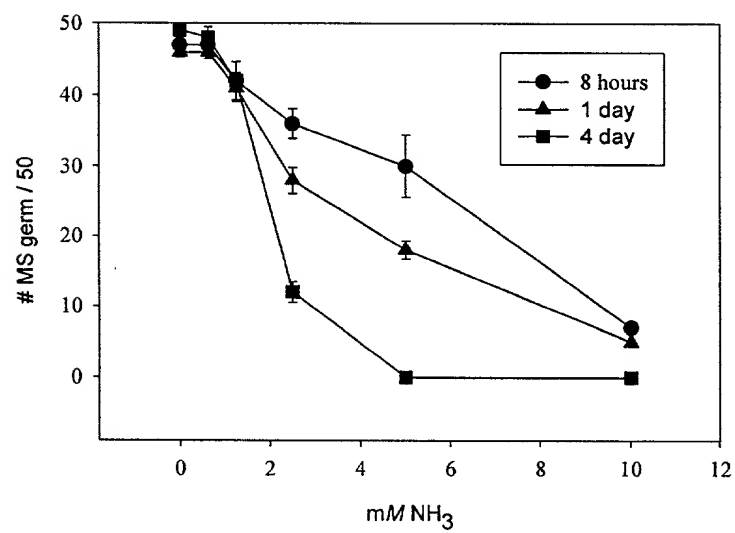


Fig. 8. Number of MS germinated out of 50 counted exposed for 8 h, 1 d and 4 d to various concentrations of NH<sub>3</sub> in glycine buffer at pH 8.6 (n=3;  $\pm$ standard error).

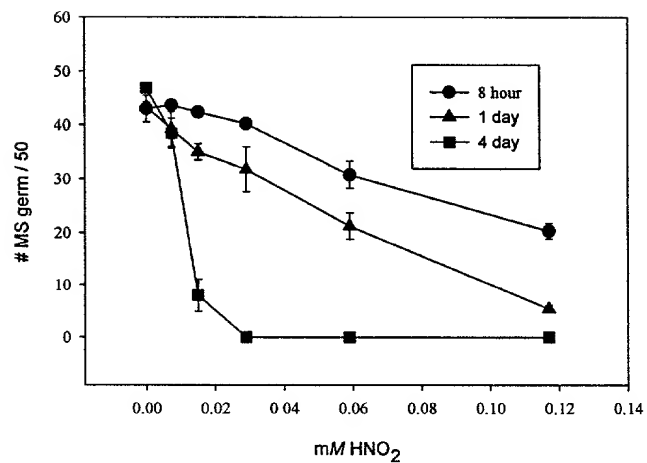


Fig. 9. Number of MS germinated out of 50 counted exposed for 8 h, 1 d and 4 d to various concentrations of HNO<sub>2</sub> in citric acid buffer at pH 5.0 (n=3;  $\pm$ standard error).

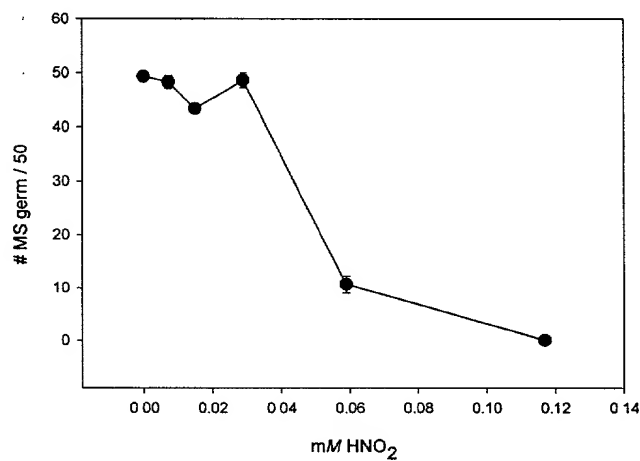


Fig. 10. Number of MS germinated out of 50 counted after suspension for 4 d to various concentrations of 30 mL HNO<sub>2</sub> in citric acid buffer at pH 5.0 in a 250 mL sealer jar (n=3;  $\pm$ standard error).

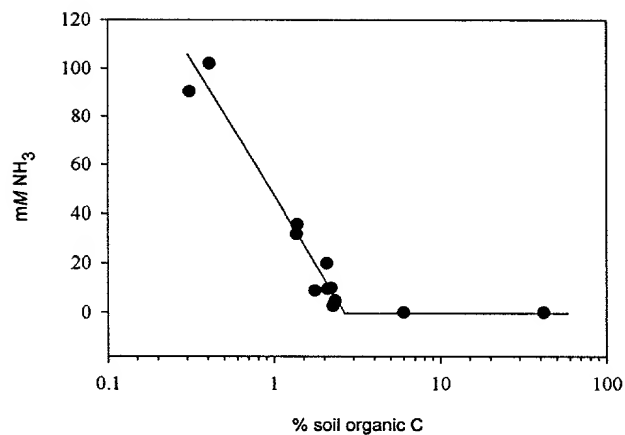


Fig. 11. Peak concentration of  $\text{NH}_3$  in soil solution measured for 12 soils amended with 2% MBM ( $\text{w w}^{-1}$ ) ( $n=3$ ;  $\pm$  standard error).

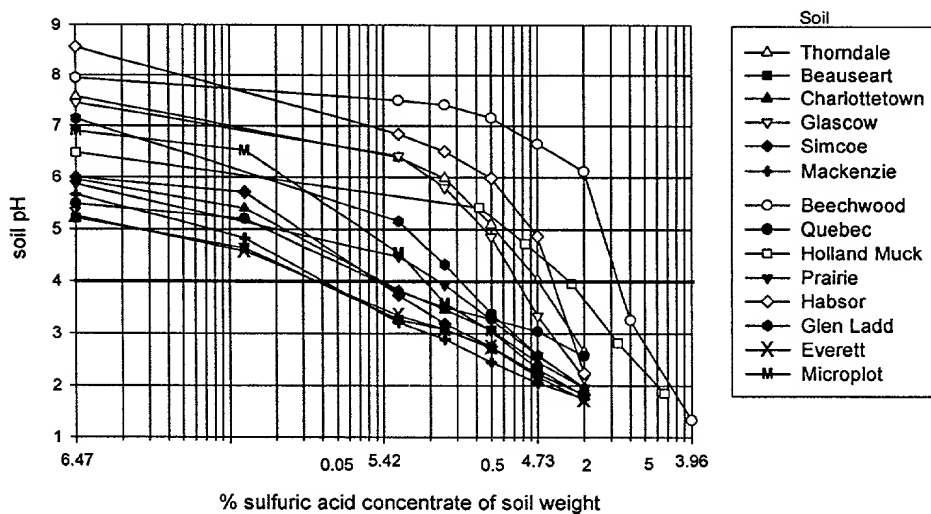


Fig. 12. Soil pH in response to addition of  $\text{H}_2\text{SO}_4$ . Soils in Group 1 are in filled symbols and Group 2 are in open symbols.



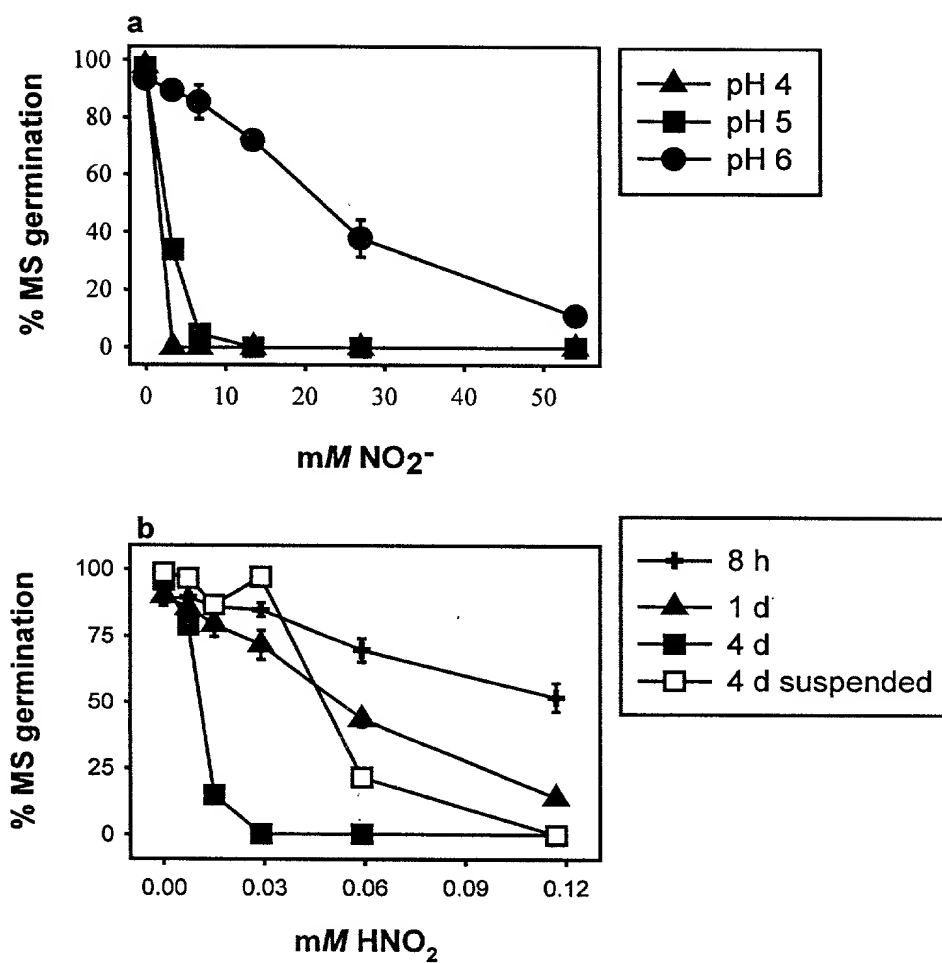


Fig. 14. Germination of *V. dahliae* MS after submergence in a 0.02 M citric acid buffered solution for a) 1 d exposure to various concentrations of NO<sub>2</sub><sup>-</sup> at a solution pH of 4, 5 or 6 and b) 8 h, 1 and 4 d exposure in, or suspended above for 4 d a solution of pH 5 containing various levels of HNO<sub>2</sub>. The concentration of HNO<sub>2</sub> was estimated based on the concentration of NaNO<sub>2</sub> and pH of the solution. Means (±1 se) of six replicates are shown.

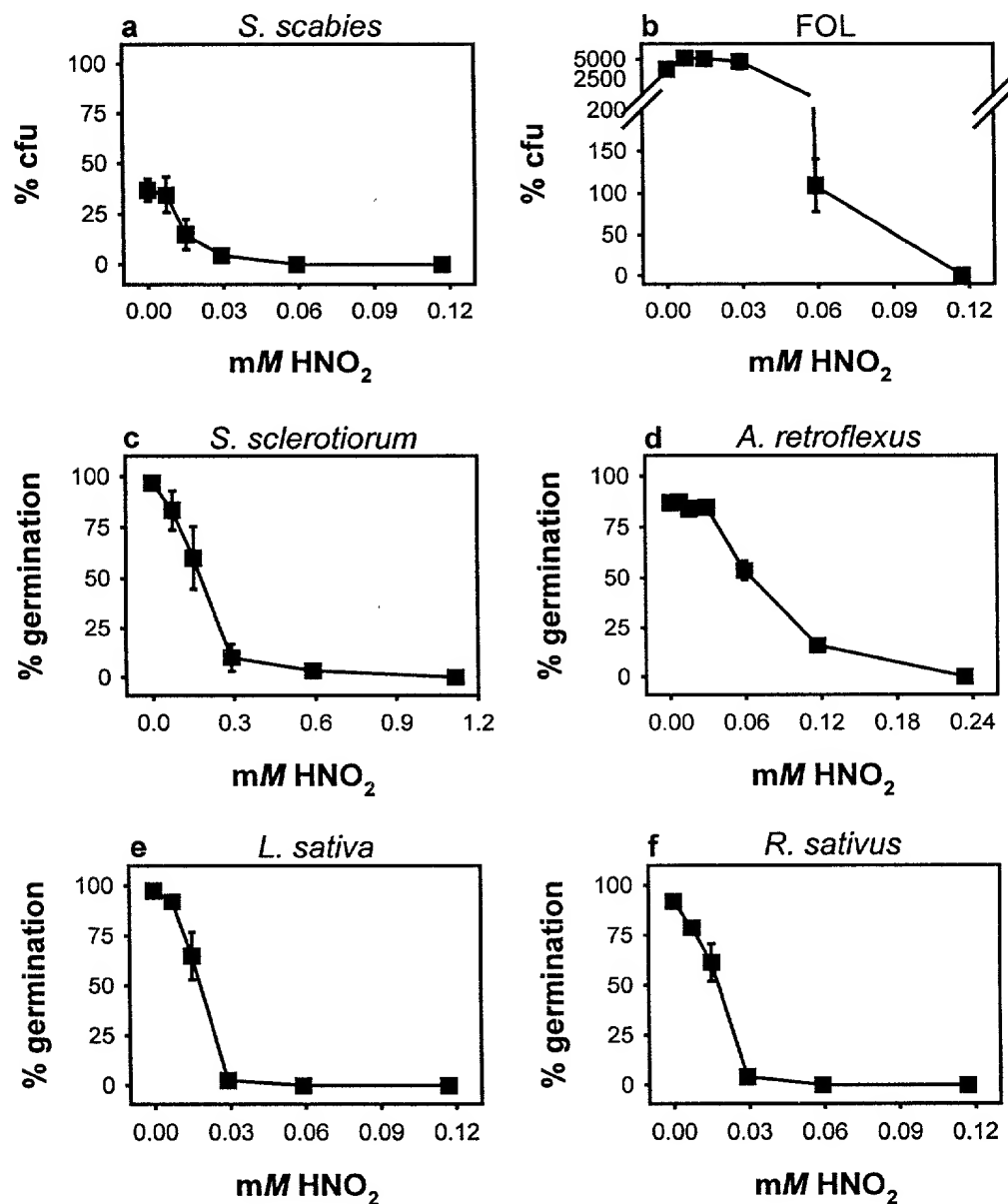


Fig. 15. Percent colony forming units (cfu) of a) spores of *Streptomyces scabies*, and b) chlamydospores of FOL, and germination of c) sclerotia of *Sclerotinia sclerotiorum*, and seeds of d) *A. retroflexus*, e) *Latuca sativa* and f) *Raphanus sativus* after submergence for 8 h, 1 and 4 d in 0.02 M citric acid buffered solutions (pH 5.0) containing various levels of  $\text{HNO}_2$ . The concentration of  $\text{HNO}_2$  was estimated based on the concentration of  $\text{NaNO}_2$  and pH of the solution. Means ( $\pm 1$  se) of six replicates are shown.